

LAUREFUCIN AND ACETYLLAUREFUCIN,  
NEW BROMO COMPOUNDS FROM LAURENCIA NIPPONICA YAMADA (1)

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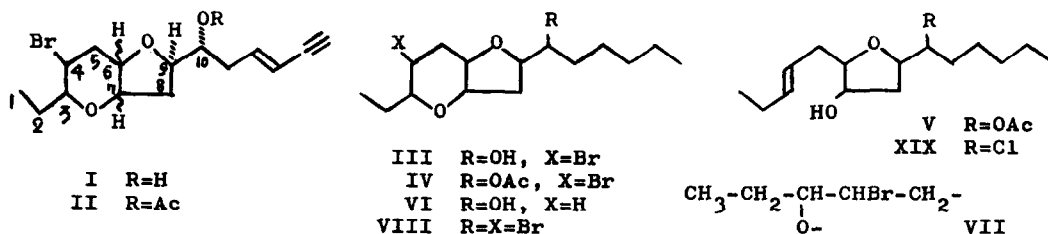
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An extensive study of the neutral essential oil from Laurencia nipponica Yamada (Rhodomelaceae; Urasozo in Japanese)(2) led to isolation of a new bromo alcohol and its acetate in 0.02 and 0.0005% yields, respectively. They are designated as laurefucin and acetyllaurefucin and are assigned structures I and II on the basis of the chemical and spectral evidence described below.

Laurefucin (I), m.p. 107-108°,  $(\alpha)_D -80^\circ$ ,  $m/e$  265 and 263 ( $M^+ - C_5H_5$ ), was analyzed for  $C_{15}H_{21}O_3Br$ . The UV ( $\lambda_{max}$  224.5 nm ( $\epsilon$  17,900),  $\lambda_{inf}$  232 (14,100)) and IR spectra ( $\nu_{max}$  3590, 3400, 3280, 2120, 1629, 1153, 1118, 1100, 1065, 1037, 1020, 965, 879  $cm^{-1}$ ) showed the presence of OH, conjugated enyne and ether functions. The NMR spectrum displayed signals due to an acetylenic proton at  $\tau$  7.18 (1H, d,  $J=2$  Hz), two vinyl protons at  $\tau$  3.76 (1H, dt,  $J=16, 7, 7$ ) and 4.45 (1H, finely splitted d,  $J=16$ ), and methyl protons at  $\tau$  9.04 (3H, sharp t,  $J=7$ ). In addition, unresolvable six-proton multiplet appeared in the region of  $\tau$  6.6-5.7, which is ascribed to protons on carbons bearing Br, OH and ether oxygen. Acetylation of I gave the corresponding acetate (II),  $C_{17}H_{23}O_4Br$ , oil,  $(\alpha)_D -126.5^\circ$ ;  $m/e$  372 and 370 ( $M^+$ ), 312 and 310 ( $M^+ - AcOH$ ), 307 and 305 ( $M^+ - C_5H_5$ );  $\lambda_{max}$  224.5 nm ( $\epsilon$  16,900),  $\lambda_{inf}$  232 (13,400);  $\nu_{max}$  3280, 2120, 1735, 1628, 1240, 1157, 1097, 1070, 1023, 960  $cm^{-1}$ ;  $\tau$  9.02 (3H, t,  $J=7$ ), 7.99 (3H, s), 7.32 (1H, d,  $J=2.5$ ), 6.55 (1H, dt,  $J=9, 9, 2$ ), 6.23 (2H, m), 5.96 (1H, br. s), 5.84 (1H, dd,  $J=7, 4$ ), 5.28 (1H, dt,  $J=5, 4, 4$ ),  $>CH-CHOAc-CH_2-$ , 4.49 (1H, f. split. d,  $J=16$ ), 3.87 (1H, dt,  $J=16, 7, 7$ ), which was identified as acetyllaurefucin isolated from the original essential

oil. Laurefucin was, therefore, shown to be a bicyclic ether having  $-\text{CH}_2-\text{CH}_3$ ,  $-\text{CH}_2-\text{CH}=\text{CH}-\text{C}\equiv\text{CH}$ ,  $>\text{CH}-\text{CHOH}-\text{CH}_2-$ , and  $>\text{CHBr}$  groupings and these were further supported by the following experiments.

I consumed 3 moles of  $\text{H}_2$  over  $\text{PtO}_2$  in ethanol to give hexahydrolaurefucin (III),  $\text{C}_{15}\text{H}_{27}\text{O}_3\text{Br}$ , m.p.  $63.0-63.5^\circ$ ,  $[\alpha]_D -53.9^\circ$ ;  $\underline{m}/\underline{e}$  318 and 316 ( $\text{M}^+ - \text{H}_2\text{O}$ );  $\nu_{\text{max}}$  3600, 3400, 1100, 1065, 1038  $\text{cm}^{-1}$ . The NMR spectrum exhibited two methyl signals at  $\tau$  9.10 (3H, br.t) and 9.06 (3H, sharp t,  $J=7$ ), while the original signals due to acetylenic and olefinic protons disappeared.

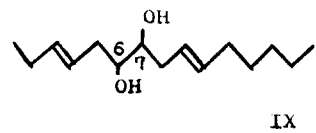


Treatment of hexahydroacetylalaurefucin (IV), acetate of III, with  $\text{Zn}-\text{AcOH}$  yielded an unsaturated acetoxy alcohol (V),  $\text{C}_{17}\text{H}_{30}\text{O}_4$  ( $\text{M}^+$  298). The IR and NMR spectra showed that V had  $-\text{CH}_2-\text{CH}=\text{CH}-\text{CH}_2-$  ( $\nu$  970  $\text{cm}^{-1}$ ;  $\tau$  4.63 and 4.50 (each 1H, sex,  $J=16, 6, 6$ ):  $\text{ABX}_2$ -type),  $>\text{CH}-\text{CHOH}-\text{CH}_2-$  ( $\nu$  3460  $\text{cm}^{-1}$ ;  $\tau$  6.56 (1H, dt,  $J=7, 7, 3$ )),  $>\text{CH}-\text{CHOAc}-\text{CH}_2-$  ( $\nu$  1745  $\text{cm}^{-1}$ ;  $\tau$  5.21 (1H, dt,  $J=7, 7, 4$ ) and 7.97 (3H, s)),  $>\text{CH}-\text{O}-\text{CH}$  ( $\nu$  1090  $\text{cm}^{-1}$ ;  $\tau$  6.10 (2H, m)), and two methyl groups ( $\tau$  9.09 (3H, br.t) and 9.04 (3H, t,  $J=7$ )). The mass spectrum showed the presence of  $\text{C}_5\text{H}_9$  group ( $\underline{m}/\underline{e}$  229 ( $\text{M}^+ - \text{C}_5\text{H}_9$ ), 151 ( $\text{M}^+ - \text{C}_5\text{H}_9 - \text{H}_2\text{O} - \text{AcOH}$ ), 69 ( $\text{C}_5\text{H}_9$ )).

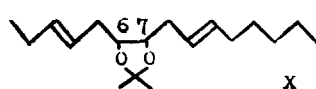
Debromination of III with Raney-Ni afforded hexahydrodebromolaurefucin (VI),  $\text{C}_{15}\text{H}_{28}\text{O}_3$  ( $\text{M}^+$  256) which showed sharp triplet ( $J=7$ ) of methyl signal at  $\tau$  9.11 and characteristic peaks due to  $\text{C}_2\text{H}_5$  group ( $\underline{m}/\underline{e}$  227 ( $\text{M}^+ - \text{C}_2\text{H}_5$ ), 209 ( $\text{M}^+ - \text{C}_2\text{H}_5 - \text{H}_2\text{O}$ )) in the NMR and mass spectra. Therefore, the partial structure VII is involved in I.

Next, III was treated with  $\text{SOBr}_2$  in ether at room temp. to give a dibromo compound (VIII),  $\text{C}_{15}\text{H}_{26}\text{O}_2\text{Br}_2$ ,  $\underline{m}/\underline{e}$  400, 398, 396 ( $\text{M}^+$ ), 371, 369, 367 ( $\text{M}^+ - \text{C}_2\text{H}_5$ ), 237 ( $\text{M}^+ - \text{Br} - \text{HBr}$ ), 221, 219 ( $\text{C}_9\text{H}_{16}\text{OBr}$ ), 179, 177 ( $\text{C}_6\text{H}_{10}\text{OBr}$ ), 139 ( $\text{C}_9\text{H}_{15}\text{O}$ ), 97 ( $\text{C}_6\text{H}_9\text{O}$ );  $\tau$  9.12 (3H, br.t), 9.04 (3H, t,  $J=7$ ), 7.94 (1H, m), 7.74 (2H, m), 7.42 (1H, quintet), 6.4-5.8 (6H, m: 2  $>\text{CHBr}$ , 4  $>\text{CH}-\text{O}-$ ). Four-proton signals in the region of  $\tau$  8.0-7.4 in the NMR spectrum could be attributed to those of

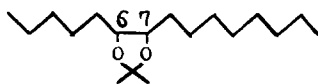
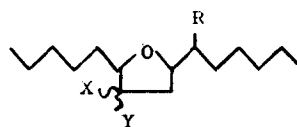
$-O-\overset{|}{\underset{|}{C}}-CH_2-\overset{|}{\underset{|}{C}}-Br$  and/or  $-O-\overset{|}{\underset{|}{C}}-CH_2-\overset{|}{\underset{|}{C}}-O-$  (2). On treatment with Zn-AcOH, VIII gave a crystalline unsaturated glycol (IX),  $C_{15}H_{28}O_2$ , m.p.  $94-95^\circ$ ,  $\nu_{max}$  3570, 1046, 977  $cm^{-1}$ ;  $\tau$  9.13 (3H, br.t), 9.03 (3H, t,  $J=7$ ), 8.74 (6H, br.s), 8.2-7.6 (10H, m), 6.46 (2H, f. split. quintet), 4.54 (4H, m). Acetonide (X) of this glycol exhibited signals due to four olefinic protons at  $\tau$  4.61 (m), two protons on carbons attached to oxygen atom at  $\tau$  6.08 (quintet), and eight allylic protons at  $\tau$  7.69. This spectrum suggested the molecule to constitute a nearly symmetrical structure. The mass spectrum of X displayed characteristic peaks at  $m/e$  265 ( $M^+ - CH_3$ ), 211 ( $M^+ - C_5H_9$ ), 169 ( $M^+ - C_8H_{15}$ ), 111 ( $C_8H_{15}$ ), 69 ( $C_5H_9$ ). Hydrogenation of X afforded a saturated acetonide (XI),  $C_{18}H_{36}O_2$ ,  $(\alpha)_D +4.2^\circ$ . The mass spectrum ( $m/e$  269 ( $M^+ - CH_3$ ), 227 ( $M^+ - C_3H_5O$ ), 213 ( $M^+ - C_5H_{11}$ ), 171 ( $M^+ - C_8H_{17}$ )) was consistent with the structure XI (3). This acetonide (XI) is very similar to XII from laurencin (2,4) and XIII from laureatin (2) in the mass spectra, but is slightly different in the IR and NMR spectra. Consequently, in view of their optical rotations, the stereochemistry at  $C_6$  and  $C_7$  of XI should be represented by R,S or S,R.



IX



X

XI  $(\alpha)_D +4.2^\circ$ XII  $(\alpha)_D +34.5^\circ (C_6R, C_7R)$ XIII  $(\alpha)_D -35.0^\circ (C_6S, C_7S)$ 

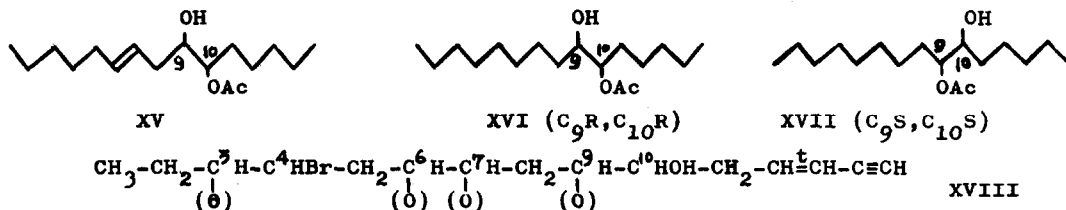
XIV R = OAc, X = Br, Y = H

XX R = Y = H, X = OH

XXI R = H, XY = O

On the other hand, hydrogenation of V followed by treatment with  $SOBr_2$  in dioxane at room temp. afforded an acetoxy bromo ether (XIV),  $C_{17}H_{31}O_3Br$ ,  $\nu_{max}$  1745, 1230, 1082, 1062, 1026  $cm^{-1}$ ;  $\tau$  7.99 (3H, s), 6.4-6.0 (3H, m), 4.87 (1H, septet);  $m/e$  223 ( $M^+ - Br - AcOH$ ), 139 ( $C_9H_{15}O$ ). Treatment of XIV with Zn-AcOH yielded an unsaturated acetoxy alcohol (XV),  $C_{17}H_{32}O_3$ ,  $\nu_{max}$  972  $cm^{-1}$ ;  $\tau$  6.54 (1H, m), 5.33 (1H, sextet), 4.66 (2H, m);  $m/e$  266 ( $M^+ - H_2O$ ). Hydrogenation of XV gave a saturated acetoxy alcohol (XVI),  $C_{17}H_{34}O_3$ ,  $(\alpha)_D +22^\circ$ ,  $\nu_{max}$  3440, 1744  $cm^{-1}$ ;  $\tau$  9.11 (6H, br.t), 7.99 (3H, s), 6.60 (1H, br.m), 5.35 (1H, f. split. quartet), which was identical with the specimen derived from laurencin ((i)  $H_2/PtO_2$  (ii) Zn-AcOH (iii)  $H_2/PtO_2$ ) in all respects. While the IR and NMR spectra of XVI were very similar as those of XVII ( $(\alpha)_D -20.3^\circ$ ) derived

from laureatin ((i)  $H_2/PtO_2$  (ii) KOH-EtOH (iii) Zn-AcOH (iv)  $Ac_2O$  (v)  $H_2/Pd-C$ ), the mass spectrum of XVI differed definitely from that of XVII. The position and stereochemistry of  $>CH-OH$  group and one ( $C_9$ ) of the carbons bearing ether oxygen in I were thus established unambiguously and, therefore, the whole skeleton of laurefucin could be represented by XVIII.



Since it is clear that laurefucin contains neither oxirane nor oxetane ring on the chemical and spectral data mentioned above, I is the most favorable structure for laurefucin.

Finally, the oxolane ring in I has been confirmed by the following experiments. XIX,  $C_{15}H_{27}O_2Cl$  ( $M^+$  276, 274), obtained from III ((i)  $SOCl_2$  (ii) Zn-AcOH), was treated with  $H_2/PtO_2$  followed by Raney-Ni to give XX,  $C_{15}H_{30}O_2$ ,  $m/e$  171 ( $M^+ - C_5H_{11}$ ), 157 ( $M^+ - C_6H_{13}$ ). XX was then oxidized with  $Ag_2O$  in benzene (reflux 48 hr) to yield the corresponding ketone XXI,  $C_{15}H_{28}O_2$  ( $M^+$  240), ( $\alpha$ ) $_D +107.1^\circ$ ;  $\tau$  9.09 (6H, br.t), 8.04 and 7.63 (each 1H,  $J_{AB}=18$ ,  $J_{AX}=13$ ,  $J_{BX}=6$ ), 6.42 (1H, br.dd), 6.02 (1H, m). This ketone XXI showed the IR absorptions,  $\nu_{max}$  1764, 1164, 1094  $cm^{-1}$ , characteristic of  $\beta$ -keto-oxolane (5).

It is concluded that the structures of laurefucin and acetyl laurefucin are represented by I and II, respectively. Laurefucin, laurencin, laureatin, and isolaureatin (2), all these compounds were isolated from Laurencia species and could be assumed to be derived biogenetically from a common precursor, pentadeca-3,9,12-trien-1-yn-6,7-diol.

#### References

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